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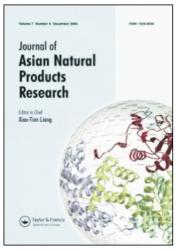
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ANDROECHIN, A NEW CHALCONE GLUCOSIDE FROM ANDROGRAPHIS ECHIOIDES

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A new chalcone glucoside, androechin, and a known flavone glucoside, echioidinin 5-O-glucoside, were isolated from the whole plant of *Andrographis echioides*. Androechin was characterized as 2,2',6'-trihydroxy-4'-methoxychalcone 2'-O- β -D-glucopyranoside by spectral and chemical studies.

Keywords: Andrographis echioides; Acanthaceae; Chalcone glucoside; Androechin

INTRODUCTION

Andrographis echioides Nees (Acanthaceae) is an erect herb found widely in the dry districts of tropical India and Sri Lanka [1] and is used in indigenous medicine as a remedy for fevers [2]. In a previous study on the hexane and Me₂CO extracts of the whole plant of A. echioides, we have reported a new flavanone, dihydroechioidinin (3) besides four known flavones [3]. Continuing our investigations on the whole plant of this species, we now report the isolation and structure elucidation of a new chalcone glucoside, designated as androechin (1) together with a known flavone glucoside, echioidinin 5-O- β -D-glucopyranoside (2) [4].

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RESULTS AND DISCUSSION

Androechin (1) was obtained as yellow crystals, m.p. 128–130°C. Its molecular formula $C_{22}H_{24}O_{10}$ was deduced from HRFABMS (m/z 471.1266 [M+Na]⁺) in conjunction with ¹³C NMR spectral data. The UV absorption maxima at 252, 310 and 365 nm [5], colour reactions [6] and positive Molisch test suggested that androechin was a chalcone glycoside. Its UV spectral maxima were unaffected by the addition of NaOAc indicating the absence of a free hydroxyl at C-4′. A downfield signal at δ 13.55 in its ¹H NMR spectrum and a bathochromic shift of 43 nm in band I of the UV spectrum with AlCl₃/HCl revealed the presence of a chelated hydroxyl in 1.

The ¹H NMR spectrum of **1** showed another phenolic hydroxyl signal at δ 10.17. A pair of AB doublets ($J = 16.0 \, \text{Hz}$) at δ 8.07 and 8.0 were consistent with *trans* olefinic protons of a chalcone moiety [7]. The methoxyl singlet at δ 3.82 was assigned to C-4′ as it showed ³J correlation with this carbon at 165.2 ppm in its HMBC spectrum (Fig. 1). A set of *meta* coupled doublets ($J = 2.3 \, \text{Hz}$) at δ 6.33 and 6.17, each integrating for one proton. were attributed to H-3′ and H-5′, as both these protons showed NOE cross peaks with methoxyl protons at C-4′ in its NOESY spectrum and HMBC correlation with C-4′ (Fig. 1). The β -carbon in C-2 unsubstituted chalcones usually resonates around 144 (\pm 2) ppm. However, in androcchin (**1**) it appeared at 137.8 ppm, which is unusually upfield, indicating the presence of C-2 oxygenation [8]. The chemical shift values of the B-ring carbons of 1 were very similar to the literature values of 2-hydroxychalcones [9] and

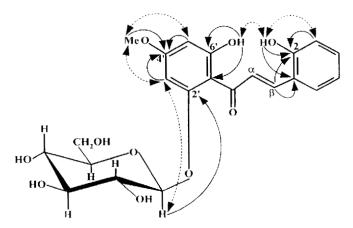


FIGURE 1 Selected HMBC (→) and NOESY (→···→) correlations observed in androcchin (1).

hence the non-chelated hydroxyl (δ 10.17) in **1** was placed at C-2. The position of C-2 hydroxyl in **1** was further evidenced by the presence of a strong NOE cross peak between the hydroxyl (δ 10.17) and C-3 proton (δ 6.91) in its NOESY spectrum, and 2J and 3J correlation with C-2 (157.0 ppm) and C-1 (121.7 ppm), respectively in its HMBC spectrum (Fig. 1). The presence of four aromatic proton signals at δ 6.91, 7.25, 6.84 and 7.79 in the 1H NMR spectrum of **1** were assigned to protons at 3, 4, 5 and 6 positions of ring B.

An anomeric proton doublet $(J = 7.0 \,\mathrm{Hz})$ at δ 5.13 in 1 indicated the presence of a sugar residue with β -configuration. Acid hydrolysis of 1 afforded D-glucose and an isomerized aglycone [10,11], identified as dihydroechioidinin (3) [3]. The presence of two phenolic hydroxyls and a glucose residue in 1 was also evidenced by the formation of an hexaccetate. The glucose residue in 1 was found to be linked to C-2' as a strong NOE was observed between H-1" and H-3' in its NOESY spectrum, and a cross peak between H-1" and C-2' (159.7 ppm) in its HMBC spectrum (Fig. 1).

Thus from the foregoing spectral and acid hydrolytic studies, androechin was characterized as 2,2',6'-trihydroxy-4'-methoxychalcone 2'-O- β -D-glucopyranoside (1).

To the best of our knowledge, isolation of androechin (1) constitutes the first report of a chalcone glycoside with 2-oxygenation in nature.

EXPERIMENTAL SECTION

General Experimental Procedures

Melting points were determined on a Kosler hot-stage apparatus and are uncorrected. UV spectra were determined in MeOH on a Shimadzu UV-240

spectrophotometer. IR spectra were obtained in KBr discs on a Perkin-Elmer 283 double beam spectrophotometer. ¹H and ¹³C NMR spectra were recorded on a Varian FT 200 and Bruker Avance 400 spectrometers using DMSO-*d*₆ and CDCl₃ with TMS as internal standard. HMBC and NOESY spectra were obtained using standard pulse sequences. FAB and HRFAB mass spectra were acquired on a 700 JEOL mass spectrometer in NBA matrix. CC was performed on Acme silica gel finer than 200 mesh (0.08 mm).

Plant Material

The whole plant of *A. echioides* Nees was collected in May, 1998 at Tirupati. Andhra Pradesh. India. A voucher specimen (DG-199) has been deposited in the herbarium of the Department of Botany, Sri Venkateswara University, Tirupati.

Extraction and Isolation

The shade dried and ground whole plant of A. echioides (3 kg) was successively extracted with n-hexane, Me₂CO and MeOH. The EtOAc soluble part of MeOH extract on purification over a silica gel column using hexane EtOAc (2:8) yielded 1 (25 mg) and 2 (20 mg).

Androechin (1)

Yeilow needles from MeOH, m.p. 128-130°C, $|\alpha|_{D}^{28}$ -22.5° (c 1.0, MeOH); UV (MeOH) λ_{max} (log ε) 252sh (3.83), 310sh (3.99), 365 (4.15); +NaOAc 252sh, 309sh, 365; +AlCl₃ 270, 408; +AlCl₃ HCl 270, 408 nm; IR (KBr) $\nu_{\rm max}$ 3341 (-OH), 1616 (C=O), 1535, 1384, 1338, 1293, 1202, 1160, 1077. 1051 cm $^{-1}$; $^{-1}$ H NMR (400 MHz, DMSO- d_6) δ 13.55 (1H, s, OH-6'), 10.17 (111, s. OH-2), 8.07 (1H, d, J = 16.0 Hz, H- α), 8.0 (1H, d, J = 16.0 Hz, H- β). 7.79 (1H, dd, J = 8.0, 1.5 Hz, H-6), 7.25 (1H, ddd, J = 8.0, 8.0, 1.5 Hz, H-4). 6.91 (111, dd, *J* = 8.0, 1.5 Hz, H-3), 6.84 (1H, ddd, *J* = 8.0, 8.0, 1.5 Hz. H-5). 6.33 (1H, d, J = 2.3 Hz, H-3'), 6.17 (1H, d. J = 2.3 Hz, H-5'), 5.13 (1H, d. $J = 7.0 \,\text{Hz}$. H-1"), 3.82 (3H, s, OMe-4'), 3.10–3.80 (6H, m, sugar protons); ¹³C NMR (75 MHz, DMSO-*d*₆) δ 192.8 (C=O), 165.3 (C-6'), 165.2 (C-4'), 159.7 (C-2'), 157.0 (C-2), 137.8 (C- β), 131.9 (C-4), 127.9 (C-6), 126.0 (C- α), 121.7 (C-1), 119.7 (C-5), 116.2 (C-3), 106.9 (C-1'), 100.5 (C-1"), 95.2 (C-5'), 93.8 (C-3'), 77.5 (C-3"), 76.8 (C-5"), 73.6 (C-2"), 69.7 (C-4"), 60.7 (C-6"), 55.6 (OMe-4'); HRFABMS m/z 471.1266 [M + Na]⁺ ($C_{22}H_{24}O_{10}$ Na requires 471.1267); FABMS m/z 471 [M + Na]⁺, 449 [M + H]⁺, 287 [M + H-162]⁺.

Echioidinin 5-O-β-D-glucopyranoside (2)

Pale yellow needles from MeOH, m.p. 245–246°C. The physical and spectral data were consistent with literature values [4].

Acetylation of 1

A mixture of compound 1 (5 mg), Ac_2O (2 ml) and C_5H_5N (1 ml) was kept at room temperature for 72 h and poured into crushed ice to yield the hexa-acetate as colourless needles (7 mg) from CHCl₃, m.p. 110°C. IR (KBr) ν_{max} 2943, 1759 (Σ =O of OAc), 1616, 1433, 1371, 1325, 1213 cm⁻¹; ¹H NMR (200 MHz, CDCl₃) δ 7.70 (1H, dd, J=8.0, 1.8 Hz, H-6), 7.45 (2H, m, H-4, 5), 7.32 (1H, d, J=15.0 Hz, H- α), 7.12 (1H, dd, J=8.0, 1.8 Hz, H-3), 6.90 (1H, d, J=15.0 Hz, H- β), 6.70 (1H, d, J=1.8 Hz, H-5'), 6.46 (1H, d, J=1.8 Hz, H-3'), 5.30 (1H, d, J=7.0 Hz, H-1"), 4.97–5.28 (4H, m, H-2", 3", 4", 5"), 4.24 (2H, m, CH₂-6"), 3.83 (3H, s, OMe-4'), 2.29 (3H, s, OAc-6'), 2.17 (3H, s, OAc-2), 1.97–2.10 (12H, m, 4 × OAc).

Acid hydrolysis of 1

A MeOH solution of 1 (10 mg) in 2 N HCl (4 ml) was heated at 100°C for 2 h. The acid hydrolysate was extracted with EtOAc and evaporated to dryness to yield a colourless solid which on crystallization from CHCl₃ afforded the aglycone as colourless needles (6 mg). It was characterized as the isomeric flavanone, dihydroechioidinin (3), m.p. 200–201°C as its physical and spectral data were very similar to the literature values [3]. The sugar in the aqueous layer was determined as D-glucose by paper chromatography.

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